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~~\u0026 Molecule Shape~~
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numbers of bonds and electron pairs? Find out by adding single, double or triple bonds and lone pairs to the central atom. Then, compare the model to real molecules!

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Molecule Shapes – Answer all
of the following questions
in your chemistry notebook!
Feel free to type your
answers in the spaces below

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if you would prefer an electronic copy. MODEL 1: Molecule Shapes Simulation PART I: ELECTRON DOMAINS Explore the Model screen of the simulation. As you explore, answer the following questions.

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bonding. STUDY. PLAY. How
does adding an atom affect~~

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the position of existing atoms or lone pairs?-it will change the molecular geometry -changes position of the lone pairs and atoms so that they are furthest apart.

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Chemistry Name _____ Block
_____ Phet Molecular Shapes

VSEPR Activity Go to

Introduction Atoms bond to
satisfy their need for more
electrons. If both atoms
have high

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electronegativities (are nonmetals), atoms will share electrons to satisfy the Octet Rule – every atom wants 8 electrons to fill the s and p orbitals in the outer energy level.

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~~...~~

It's All in the Shape:
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~~Interactive Simulations~~

Molecule Shapes- inquiry:

Description This was used in class or as homework depending on the schedule.

Learning Goals: Students will be able to: •Identify substances to which

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“Molecular geometry”
applies. •Name molecule and
electron geometries for
basic molecules. •Explain
the model being used to
predict molecule geometry.

~~Molecule Shapes inquiry~~

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~~PhET Contribution~~

Molecule Shapes simulation.

2. On the Model tab have a play to familiarise yourself with the program. Things to try include: a. Clicking and holding the mouse button then moving the cursor in

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order to rotate the molecule
b. Checking the show bond
angles box c. Checking the
molecule geometry and
electron geometry boxes d.

~~MODELLING MOLECULAR SHAPES~~
~~Ms. Suchy's science site~~

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When is a molecule polar?
Change the electronegativity of atoms in a molecule to see how it affects polarity. See how the molecule behaves in an electric field. Change the bond angle to see how shape affects polarity.

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~~Molecule Polarity — Polarity
| Electronegativity | Bonds
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Molecular Geometry Flash
Cards: Description The flash
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be used with the PhET

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simulation “Molecule Shapes.” This activity guides students to build the 13 molecular geometries taught in an introductory chemistry course, and the end product doubles as a study tool and quick

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reference for students.

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