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is made up of atoms, elements, compounds,
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Composition 1. 100 washers 0.110 g 1
washer = 11.0 g (assuming 100 washers is
exact) $100. \text{ g}$ 1 washer $0.110 \text{ g} = 909$
washers 2. $500. \text{ g}$ 1 cork $1.63 \text{ g} = 306.7 =$
 307 corks $500. \text{ g}$ 1 stopper $4.31 \text{ g} = 116$
stoppers 1 kg (1000 g) of corks contains
(1000 g

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6. (c) 7. (d) 8. (b) 9. (a) 10. (c) NCERT Solutions for Class 11 Chemistry Chapter 6 HOTS Questions. Question 1. Why standard entropy of an elementary substance is not zero whereas the standard enthalpy of formation is taken as zero? Answer: A substance has a perfectly ordered arrangement only at absolute zero. Hence, entropy is zero only at ...

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Thermodynamics is the part of Chemistry which deals with exchange of energy in the form of heat and temperature. William Thomson coined this term in 1794 by

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combining two Greek words *Thermes* which is heat and *Dynamikos* meaning powerful.

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Chapter 6 Equilibrium Chemistry 211 the reaction we monitor the mass of Ca^{2+} remaining in solution and the mass of CaCO_3 that precipitates, the result looks something like Figure 6.1. At the start of the reaction the mass of Ca^{2+} decreases and the mass of CaCO_3 increases. Eventually the reaction reaches a point

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after which there
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Chapter 6

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